



Ti(III)-mediated radical cyclization of β -aminoacrylate containing epoxy alcohol moieties: synthesis of highly substituted azacycles [☆]

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ABSTRACT

Ti(III)-mediated radical cyclization of β -aminoacrylate containing 2,3-epoxy alcohol moieties led to the formation of highly substituted piperidine and pyrrolidine rings. The pyrrolidine ring system was then transformed into an indolizidine framework present in many natural products.

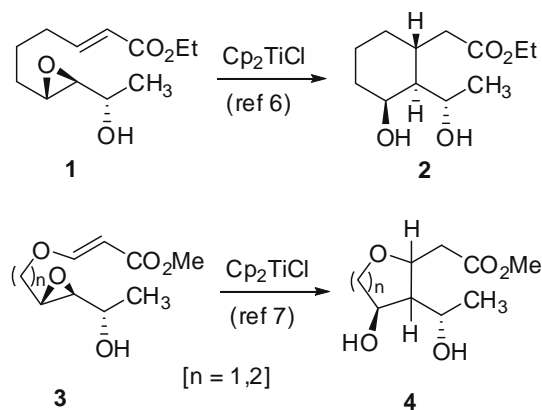
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Piperidine, pyrrolidine, and indolizidine/quinolizidine are important structural scaffolds of several natural products.¹ In the literature, radical cyclization of β -alkoxyacrylates and β -aminoacrylates has been extensively used as versatile tools for the construction of oxacyclic^{2,3} and azacyclic⁴ rings with the latter having applications in the synthesis of many alkaloids. Recently, we have reported that radicals formed during the opening of 2,3-epoxy alcohols **1** and **3** with $\text{Cp}_2\text{Ti(III)Cl}$ ⁵ could be trapped intramolecularly by a suitably positioned α,β -unsaturated ester moiety in the same molecule giving rise to a cyclohexane ring system **2**,⁶ tetrahydrofurans, and tetrahydropyrans **4** (see Scheme 1).⁷

Focusing on our work on the synthesis of carbocycles, oxacycles, and azacycles via Ti(III)-mediated radical cyclization reactions, we wish to report here the cyclization reaction of β -aminoacrylates through epoxide opening followed by 5-exo and 6-exo cyclizations. The details of the process are outlined in Schemes 2–4. Scheme 2 describes the synthesis of a highly substituted piperidine moiety. The synthesis started from the commercially available compound **5**. Tosylation of **5** with tosyl chloride followed by treatment with methyl propiolate in the presence of *N*-methylmorpholine (NMM) gave the ' β -aminoacrylate' intermediate **6**.⁸ Cleavage of the acetal **6** with formic acid followed by Wittig olefination with

stabilized ylide $\text{Ph}_3\text{P}=\text{CHCOCH}_3$ led to an α,β -unsaturated keto compound **7**.

A Luche reduction⁹ of **7** followed by a Sharpless kinetic resolution¹⁰ of the resultant racemic allylic alcohol afforded chiral epoxy alcohol **8** >92% ee as determined using the Mosher ester method¹¹ in 45% yield. With this epoxide in our hand, we turned our attention to carry out the crucial epoxide ring opening reaction followed by cyclization. Accordingly, when epoxy alcohol **8** was treated with $\text{Cp}_2\text{Ti(III)Cl}$, generated in situ from Cp_2TiCl_2 , Zn dust and freshly fused ZnCl_2 , it underwent epoxide opening at the C-2 position from

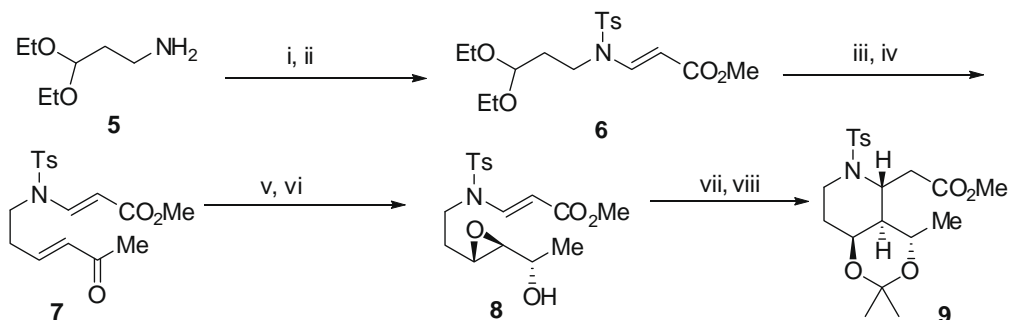


Scheme 1.

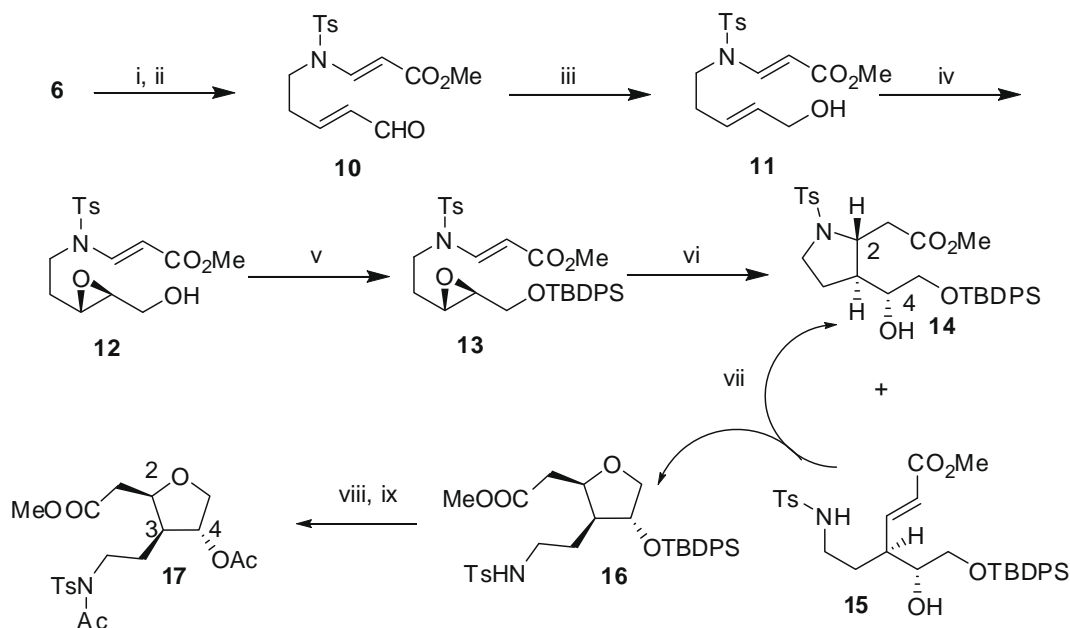
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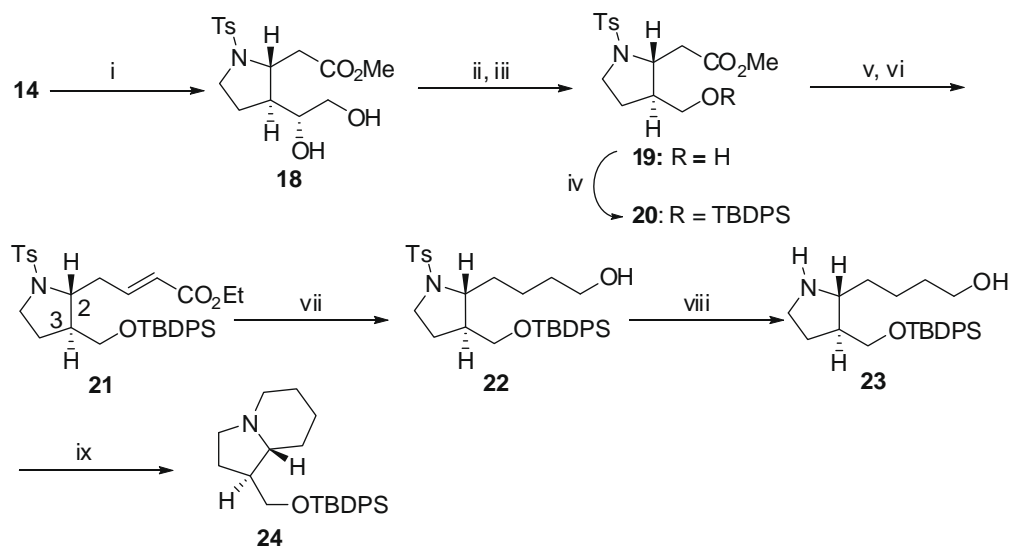
E-mail address: chakraborty@cdri.res.in (T.K. Chakraborty).



Scheme 2. Reagents and conditions. (i) TsCl, Et₃N, DMAP (cat.), CH₂Cl₂, 0 °C to rt, 2 h; (ii) methyl propiolate, NMM, CH₂Cl₂, rt, 1 h, 76% over two steps; (iii) 20% HCO₂H, pentane, 0 °C, 0.5 h; (iv) Ph₃P=CHCOCH₃, CH₂Cl₂, rt, 8 h, 85% over two steps; (v) NaBH₄, CeCl₃, MeOH, 0 °C, 15 min; (vi) L-(+)-DIPT, Ti(OⁱPr)₄, TBHP, MS (4 Å), CH₂Cl₂, –20 °C, 0.5 h, 45% over two steps; (vii) Cp₂TiCl₂, ZnCl₂, Zn, THF, –20 °C to rt, 8 h; (viii) 2,2-dimethoxypropane, CSA (cat.), CH₂Cl₂, 2 h, 40% in two steps.



Scheme 3. Reagents and conditions. (i) 20% HCO₂H, pentane, 0 °C, 0.5 h; (ii) Ph₃P=CHCHO, C₆H₆, reflux, 6 h, 60% in two steps; (iii) NaBH₄, CeCl₃, MeOH, rt, 24 h, 55%; (iv) L-(+)-DIPT, Ti(OⁱPr)₄, TBHP, MS (4 Å), CH₂Cl₂, –20 °C, 2 h, 85%; (v) TBDPSCI, Et₃N, CH₂Cl₂, DMAP (cat.), 0 °C to rt, 4 h, 95%; (vi) Cp₂TiCl₂, ZnCl₂, Zn, THF, –20 °C to rt, 6 h; (vii) **15**, K₂CO₃, MeOH, 0 °C, 2 h, 69% (combined yield) over two steps; (viii) TBAF, THF, 0 °C to rt, 2 h, 85%; (ix) Ac₂O, Et₃N, DMAP, CH₂Cl₂, 0 °C to rt, 0.5 h, 90%.



Scheme 4. Reagents and conditions. (i) TBAF, THF, 0 °C, 1 h; (ii) NaIO₄, THF/H₂O (1:1) 0 °C, 15 min; (iii) NaBH₄, MeOH, rt, 10 min; (iv) TBDPSCI, Et₃N, CH₂Cl₂, DMAP (cat.), 0 °C to rt, 4 h, 70% over four steps; (v) DIBAL-H, CH₂Cl₂, –78 °C, 15 min; (vi) Ph₃P=CHCO₂Et, CH₂Cl₂, rt, 8 h, 80% in two steps; (vii) LiBH₄, THF/H₂O (20:1), 0 °C to rt, 24 h, quantitative; (viii) Na⁺C₁₀H₈[–] DME, –60 °C, 10 min, 85%; (ix) Ph₃P, CBr₄, Et₃N, CH₂Cl₂, 24 h, 60%.

the hydroxy side¹² and gave a radical intermediate that underwent facile intramolecular trapping by the acrylate moiety leading to the formation of the six-membered piperidine as the only isolable product along with some unidentified complex mixture of compounds. Next, the resulting diol was protected as an acetonide to furnish the bicyclic compound **9** as a white crystalline solid.¹³

The absolute stereochemistry of **9** was established unequivocally from its single-crystal X-ray analysis¹⁴, which confirmed the assigned structure (Fig. 1).

Next, we wanted to test this reaction in a substrate containing primary epoxy alcohol. For this, we started from compound **6** as shown in Scheme 3. Cleavage of the acetal protection with formic acid followed by the Wittig reaction of the resulting aldehyde with $\text{Ph}_3\text{P}=\text{CHCHO}$ in refluxing benzene furnished the α,β -unsaturated aldehyde **10** in 60% yield over two steps.

The Luche reduction⁹ of **10** provided the allylic alcohol **11**, which was subjected to Sharpless asymmetric epoxidation¹⁰ using L -(+)-DIPT to furnish chiral epoxy alcohol **12**. However, treatment of the primary epoxy alcohol with $\text{Cp}_2\text{Ti(III)Cl}$ gave only an allylic alcohol¹⁵ and no cyclization product was obtained. The primary hydroxyl group was then protected as a silyl ether and when this epoxide **13** was treated with Ti(III) reagent, it opened the epoxy ring at the C-3 position and the radical at C-3 was trapped intramolecularly by the acrylate moiety furnishing a mixture of the desired cyclized pyrrolidine **14**¹⁶ (minor product, 20%) and a ring opened acyclic product **15** (major one, 70%), which was probably formed by in situ opening of the pyrrolidine **14**. Both **14** and **15** were found to have isomeric products at C3-H in a 4:1 ratio. Compound **15** could, however, be transformed back into the same pyrrolidine **14** in 70% yield on treatment with K_2CO_3 in methanol taking its overall yield to 69%. In this process, we also obtained another highly substituted tetrahydrofuran **16** (~4:1 diastereomeric mixture) in 20% yield from **15**. To know the absolute stereochemistry of **14** (major isomer), we first assigned the stereochemistry of **17**, which was obtained from **16** in two steps. During the course of radical-mediated epoxide opening and subsequent base-catalyzed cyclization, the absolute stereochemistry at C-4 of **14** was retained

as *R* as it was in the chiral epoxide **12**. The C-5 protons decoupled ¹H NMR spectrum of **17** showed a doublet ($J = 1.62$ Hz) at 5.03 ppm for C4-H signal indicating that the C3-H and C4-H had a *trans*-relationship and that the absolute stereochemistry of C-3 in **17**, and hence in **14**, was *S*. The absolute stereochemistry of C-2 in **14** was established at a later stage.

Next, we wanted to transform the pyrrolidine moiety to an indolizidine frame work, which is a very important building block for many natural products.^{1e-k} For the synthesis of the indolizidine frame work, shown in Scheme 4, we started from **14** which was treated with TBAF to provide diol **18**. Further oxidative cleavage of the resulting diol with NaIO_4 gave an aldehyde that was treated with NaBH_4 to form primary alcohol **19**. The protection of the primary alcohol of **19** as a TBDPS ether gave **20** as a single isomer after removing the minor isomer via silica gel column chromatography. The treatment of **20** with 1 equiv of DIBAL-H followed by Wittig olefination with stabilized ylide $\text{Ph}_3\text{P}=\text{CHCO}_2\text{Et}$ gave α,β -unsaturated ester compound **21**¹⁷ as a white crystalline compound. The stereochemistry of **21** was determined by the ³J values of the C2-H proton. It appeared as a ddd at 3.62 ppm with coupling constants of 7.8, 3.7, and 3.5 Hz. One of the $\text{CH}_2\text{-CH}=\text{CH-CO}_2\text{Et}$ protons appeared as a ddd at 2.67 ppm with coupling constants 14.5, 7.4, and 3.7 Hz. The other one appeared as a td at 2.58 ppm with coupling constants 14.5 and 7.8 Hz. So the coupling constant between C2-H and C3-H is 3.5 Hz, which indicates that the relationship between C2-H and C3-H was *trans*. The absolute stereochemistry of **21** was, finally, unequivocally established from the single-crystal X-ray analysis¹⁸ which clearly showed the assigned structure (Fig. 2). Consequently, it also proved that the absolute stereochemistry at C-2 in **14** was *R*.

Next, the reduction of **21** with LiBH_4 gave saturated primary alcohol **22**, which on treatment with sodium naphthalenide¹⁹ provided the desotylated product **23**. The transformation of primary alcohol to the corresponding alkyl bromide followed by cyclization²⁰ gave the desired indolizidine framework **24**. The spectral and analytical data of **24**²¹ were in good agreement with those reported in the literature.

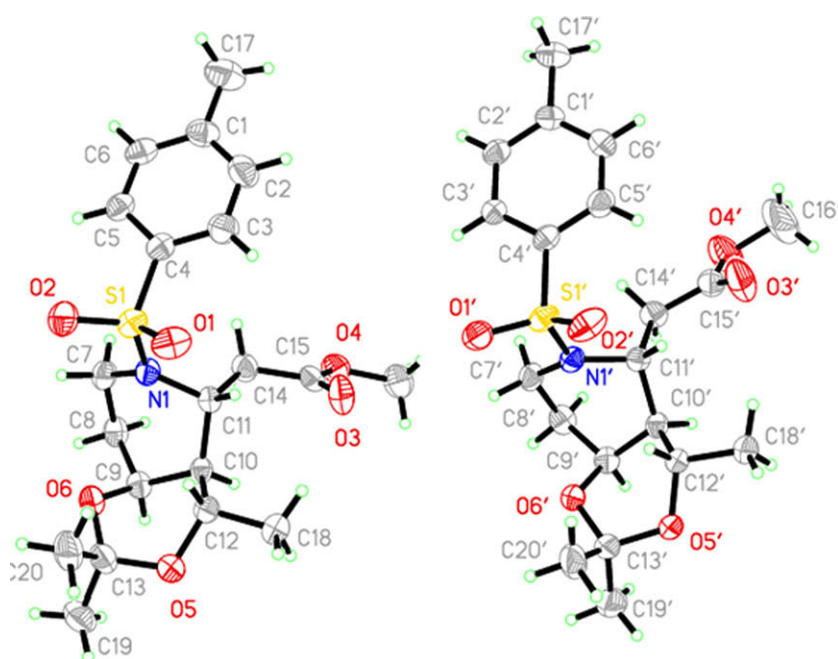


Figure 1. X-ray crystal structure of **9**. Perspective view of the two independent molecules showing the atom-numbering schemes. Displacement ellipsoids are drawn at the 30% probability level, and H atoms are shown as small spheres of arbitrary radii.

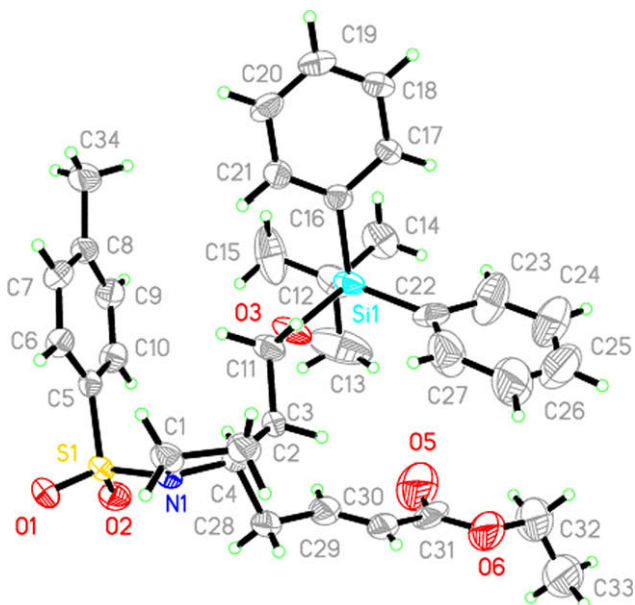


Figure 2. X-ray crystal structure of **21**. Displacement ellipsoids are drawn at 30% probability level, and H atoms are shown as small spheres of arbitrary radii.

In conclusion, we have demonstrated the Ti(III)-mediated radical cyclization of 'β-aminoacrylate' containing 2,3-epoxy alcohols, and this method can be extended to the synthesis of many natural products containing piperidine, pyrrolidine, and indolizidine/quinolizidine moieties.

Acknowledgments

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- Analytical and spectral data of compound 9*: *R*_f = 0.4 (silica gel, 30% EtOAc in hexane); [α]_D²⁵ +25.8 (c 0.53 in CHCl₃); IR (neat): ν_{max} 2985, 2934, 1735, 1332, 1154 cm⁻¹; ¹H NMR (200 MHz, CDCl₃): δ 7.71 (d, *J* = 8.0 Hz, 2H), 7.28 (d, *J* = 8.0 Hz, 2H), 4.17–3.98 (m, 2H), 3.65 (s, 3H), 3.58 (m, 1H), 3.37 (m, 1H), 3.18 (td, *J* = 11.6, 5.1 Hz, 1H), 2.54–2.46 (m, 2H), 2.43 (s, 3H), 2.16–1.81 (m, 2H), 1.60 (dd, *J* = 9.4, 6.5 Hz, 1H), 1.24 (s, 3H), 1.15 (d, *J* = 6.5 Hz, 3H), 1.14 (s, 3H); ¹³C NMR (75 MHz, CDCl₃): δ 170.7, 143.4, 137.0, 129.6, 127.0, 99.3, 64.8, 62.9, 51.8, 48.9, 46.4, 39.8, 37.8, 26.5, 26.4, 24.7, 21.4, 18.9; MS (ESI): *m/z* (%) 412 (15) [M+H]⁺, 434 (35) [M+Na]⁺; HRMS (ESI): calcd for C₂₀H₂₉NO₆NaS [M+Na]⁺ 434.1613, found 434.1609.
- X-ray Crystal data for compound 9*: Crystal data, C₂₀H₂₉NO₆S, *M* = 411.5, orthorhombic, space group *P*2₁2₁2₁, *a* = 8.2570(6) Å, *b* = 18.0755(14) Å, *c* = 28.902(2) Å, *V* = 4313.6(5) Å³, *d*_{calcd} = 1.267 Mg m⁻³. Data were collected at room temperature using a Bruker Smart Apex CCD diffractometer with graphite-monochromated MoKα radiation (λ = 0.71073 Å) with ω-scan method.²² Preliminary lattice parameters and orientation matrices were obtained from four sets of frames. Unit cell dimensions were determined from the setting angles of 9359 reflections for compound **9**. Integration and scaling of intensity data were accomplished using SAINT program.²² The structure was solved by Direct Methods using SHELXS97²³ and refinement was carried out by full-matrix least-squares technique using SHELXL97.²³ All the hydrogen atoms were positioned geometrically and were treated as riding on their parent carbon atoms, with C–H distance of 0.93–0.98 Å and an O–H = 0.82 Å, with *U*_{iso}(H) = 1.2*U*_{eq}(C) or 1.5*U*_{eq}(methyl C and O). The structure was refined with *R*₁ = 0.0373, *wR*₂ = 0.0972 for 986 reflections with *I* > 2σ(*I*). Crystallographic data has been deposited for compound **9** with the Cambridge Crystallographic Data Centre [CCDC No. 696654]. Copies of the data can be obtained free of charge at www.ccdc.cam.ac.uk/conts/retrieving.html [or from the Cambridge Crystallographic Data Centre (CCDC), 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44(0) 1223 336 033; email: deposit@ccdc.cam.ac.uk].
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- Analytical and spectral data of compound 14* (major isomer): *R*_f = 0.6 (silica gel, 30% EtOAc in hexane); IR (neat): ν_{max} 2932, 1735, 1431, 1341, 1159, 1104 cm⁻¹; ¹H NMR (200 MHz, CDCl₃): δ 7.77–7.19 (m, 14H), 3.71 (m, 1H), 3.57 (s, 3H), 3.54–3.28 (m, 2H), 3.24–2.99 (m, 3H), 2.93 (dd, *J* = 16.1, 3.6 Hz, 1H), 2.53 (dd, *J* = 16.1, 8.8 Hz, 1H), 2.38 (s, 3H), 2.01 (m, 1H), 1.80–1.63 (m, 2H), 1.03 (s, 9H); ¹³C NMR (75 MHz, CDCl₃): δ 171.4, 143.5, 135.4, 134.1, 132.7, 129.9, 129.6, 127.8, 127.6, 70.2, 66.2, 58.6, 51.5, 48.0, 46.7, 40.8, 26.7, 24.3, 21.5, 19.1; MS (ESI): *m/z* (%) 596 (45) [M+H]⁺, 618 (30) [M+Na]⁺; HRMS (ESI): calcd for C₃₂H₄₁NO₆NaSiS [M+Na]⁺ 618.2321, found 618.2300.
- Analytical and spectral data of compound 21*: *R*_f = 0.5 (silica gel, 30% EtOAc in hexane); [α]_D²⁵ –22.9 (c 0.63 in CHCl₃); IR (neat): ν_{max} 2937, 2862, 1718, 1344, 1161, 1103 cm⁻¹; ¹H NMR (500 MHz, CDCl₃): δ 7.65 (d, *J* = 8.2 Hz, 2H), 7.49–7.38 (m, 6H), 7.36–7.30 (m, 4H), 7.17 (d, *J* = 8.2 Hz, 2H), 6.90 (td, *J* = 15.6, 7.5 Hz, 1H), 5.85 (d, *J* = 15.6 Hz, 1H), 4.16 (q, *J* = 6.7 Hz, 2H), 3.62 (ddd, *J* = 7.8, 3.7, 3.5 Hz, 1H), 3.37 (m, 1H), 3.05 (dd, *J* = 9.7, 8.2, 7.4 Hz, 1H), 2.93 (d, *J* = 7.4 Hz, 2H), 2.67 (ddd, *J* = 14.5, 7.4, 3.7 Hz, 1H), 2.58 (td, *J* = 7.8, 14.5 Hz, 1H), 2.36 (s, 3H), 2.04 (m, 1H), 1.82 (m, 1H), 1.43 (m, 1H), 1.28 (t, *J* = 7 Hz, 3H), 0.98 (s, 9H); ¹³C NMR (75 MHz, CDCl₃): δ 166.1, 144.3, 143.4, 135.4, 134.0, 133.1, 129.8, 129.6, 127.7, 127.4, 124.3, 63.4, 60.7, 60.3, 47.5, 45.1, 38.9, 26.7, 25.8, 21.5, 19.0, 14.2; MS (ESI): *m/z* (%) 606 (15) [M+H]⁺, 623 (100) [M+NH₄]⁺; HRMS (ESI): calcd for C₃₄H₄₃NO₅NaSiS [M+Na]⁺ 628.2528, found 628.2498.
- (a) *X-ray Crystal data for compound 21*: Crystal data, C₃₄H₄₃NO₅SiS, *M* = 605.84, monoclinic, space group *P*2₁, *a* = 10.2220(7) Å, *b* = 8.2252(6) Å, *c* = 19.9503(14) Å, β = 97.939(1)°, *V* = 1661.3(2) Å³, *d*_{calcd} = 1.211 Mg m⁻³. Data were collected at room temperature using a Bruker Smart Apex CCD diffractometer with graphite-monochromated MoKα radiation (λ = 0.71073 Å) with ω-scan method.²² Preliminary lattice parameters and orientation matrices were obtained from four sets of frames. Unit cell dimensions were determined from the setting angles of 5835 reflections for

compound **21**. Integration and scaling of intensity data were accomplished using the SAINT program.²² The structure was solved by Direct Methods using SHELXS97,²³ and refinement was carried out by full-matrix least-squares technique using SHELXL97.²³ The side chain atoms C30/C31/C32/C33/O6 are disordered over two sites with occupancies of 0.711(14) and 0.289(14). The geometries of the disordered atoms were refined with distance constraints. The displacement parameters of the disordered atoms were restrained. All the hydrogen atoms were positioned geometrically and were treated as riding on their parent carbon atoms, with C–H distance of 0.93–0.98 Å and an O–H = 0.82 Å, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ or $1.5U_{\text{eq}}$ (methyl C and O). The structure was refined with $R_1 = 0.0672$, $wR_2 = 0.1777$ for 5199 reflections with $I > 2\sigma(I)$. The structure is shown in Figure 2. The absolute stereochemistry was confirmed by refinement of the absolute structure parameters {Flack parameter = 0.08(13)}. Crystallographic data have been deposited for compound **21** with the Cambridge Crystallographic Data Center [CCDC No. 696653]. Copies of the data can be obtained free of charge at www.ccdc.cam.ac.uk/conts/retrieving.html [or from the Cambridge

- Crystallographic Data Center (CCDC), 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44(0) 1223 336 033; email: deposit@ccdc.cam.ac.uk.]; (b) Flack, H. D.; Bernardinelli, G. *J. Appl. Crystallogr.* **2000**, 33, 1143–1148.
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 - Analytical and spectral data of compound 24*: $R_f = 0.3$ (silica gel, 10% MeOH in CHCl_3); $[\alpha]_{\text{D}}^{25} +31.1$ (c 0.37 in CHCl_3); IR (neat): ν_{max} 2930, 2858, 1464, 1430, 1108 cm^{-1} ; $^1\text{H NMR}$ (200 MHz, CDCl_3): δ 7.74–7.30 (m, 10H), 3.63 (d, $J = 4.4$ Hz, 2H), 3.23–3.04 (m, 2H), 2.30–1.14 (m, 12H), 1.05 (s, 9H); $^{13}\text{C NMR}$ (75 MHz, CDCl_3): δ 135.6, 133.6, 129.6, 127.6, 67.4, 64.5, 53.3, 52.9, 45.1, 29.6, 29.3, 26.8, 24.4, 23.9, 19.2; MS (ESI): m/z (%) 394 (100) $[\text{M}+\text{H}]^+$; HRMS (ESI): calcd for $\text{C}_{25}\text{H}_{36}\text{NOSi}$ $[\text{M}+\text{H}]^+$ 394.2566, found 394.2549.
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